Hey and Associates, Inc.

DREW CREEK MONITORING PROJECT FINAL REPORT

FOX LAKE, DODGE COUNTY, WISCONSIN



Prepared for:

Fox Lake Inland Lake Protection and Rehabilitation District W10543 HWY F Fox Lake WI 53933

December 1, 2009

PN: 08167

ntroduction	2
Sampling Methods	.3
Summary of UW-Milwaukee 2006 Sampling	5
Results of Hey Sampling	6
Stream Flow	
Total Suspended Solids	
Phosphorus	
Nitrogen	
Fecal Bacteria	
Sampling by Fox Lake Correctional Institution	11
Conclusions	14
Recommendations for Future Action	14

INTRODUCTION

Fox Lake is a 1,022-hectare (2,625-acre) lake located in northwestern Dodge County. In the 1980's and 1990's, Fox Lake experienced a rapid shift in water quality from a clearwater lake to one characterized by poor-water transparency, increased algae populations, loss of aquatic macrophytes, loss of wetland fringe, and declining sports fishery. In the mid 1990's, the Fox Lake Inland Lake Protection and Rehabilitation District (FLILPRD), in partnership with the Wisconsin Department of Natural Resources (WDNR) began implementation of a long-range management project to shift the lake back into a clear-water state. In 1995 a long-range management strategy for Fox Lake was developed by an advisory committee that included FLILPRD, WDNR, Dodge County, University of Wisconsin-Extension, Town of Fox Lake, City of Fox Lake, and civic and sportsman groups. The project management strategy is outlined in a report tilted, *Long Range Planning Strategy for the Rehabilitation of Fox Lake, Dodge County* (R. A. Smith and Associates, Inc. 1998).

To deal with the complex water quality problems at Fox Lake, the planning and rehabilitation process was broken down into the following components:

- 1. Watershed management to reduce sediment and nutrient inputs
- 2. Shoreline stabilization to reduce erosion
- 3. Aquatic plant management to restore rooted aquatic vegetation
- 4. Fishery Management (bio-manipulation to reduce rough fish and increase top predators)
- 5. Lake use management to protect sensitive areas
- 6. Public education

In 2005 and 2006 the University of Wisconsin and Hey and Associates, Inc. conducted an intensive lake and watershed monitoring program to evaluate the success of the above management strategy. The results of the monitoring are summarized in a report titled: *Fox Lake Management Strategy Evaluation and Recommendations for Future Action – 2008*, (Hey and Associates, Inc. and UW-Milwaukee, 2008). The monitoring documented that high levels of nitrogen and phosphorus were entering the lake from the lake's three tributaries.

The purpose of this project was to collect additional data on sources of nitrogen, phosphorus and sediment entering Fox Lake from the Drew Creek watershed. The work is a follow-up to sampling in 2005 and 2006 conducted by the University of Wisconsin-Milwaukee. The goal of the project is to narrow down which watershed activities, such as feedlots, animal waste storage and spreading, wastewater treatment, and tillage practices on specific properties are contributing to the high concentration of nitrogen and phosphorus being experienced in the previous sampling.

SAMPLING METHODS

Sampling was conducted at six sites illustrated on Figure 1. Samples were collected on four dates in the fall of 2008 and summer of 2009. Samples were analyzed for the following parameters:

- TOTAL KJELDAHL NITROGEN
- NITRATE PLUS NITRITE-NITROGEN
- TOTAL PHOSPHORUS
- DISSOLVED PHOSPHORUS
- TOTAL SUSPENDED SOLIDS (SEDIMENT)
- FECAL COLIFORM (MFFCC) (BACTERIA)
- STREAM FLOW

All sampling was conducted using the methods outlined in:

- Edwards, T.K., and G.D. Glysson. 1999. *Field Methods for Measurement of Fluvial Sediment, Book 3, Chapter C2*. Techniques of Water-Resources Investigations of the United States Geological Survey, U.S. Government Printing Office, Washington, DC.
- Shelton, L. R., 1994. *Field Guide for Collecting and Processing Stream Water Samples for the National Water-Quality Assessment Program*, Open-File Report 94-455, United States Geological Survey, Sacramento, California.
- United States Geological Survey (USGS). 2005. Techniques of Water Resources Investigations Reports. Book 3: Applications of hydraulics, Section A: Surface-water techniques. (21 chapters). United States Department of Interior, U.S. Geological Survey. Washington D.C. <u>http://water.usgs.gov/pubs/twri/</u>.

All water quality and bacterial samples were iced upon collection and transported by cooler to the Wisconsin State Laboratory of Hygiene in Madison Wisconsin for analysis.

Flow velocities were measured using a Marsh McBerny FlowMate® flow meter.

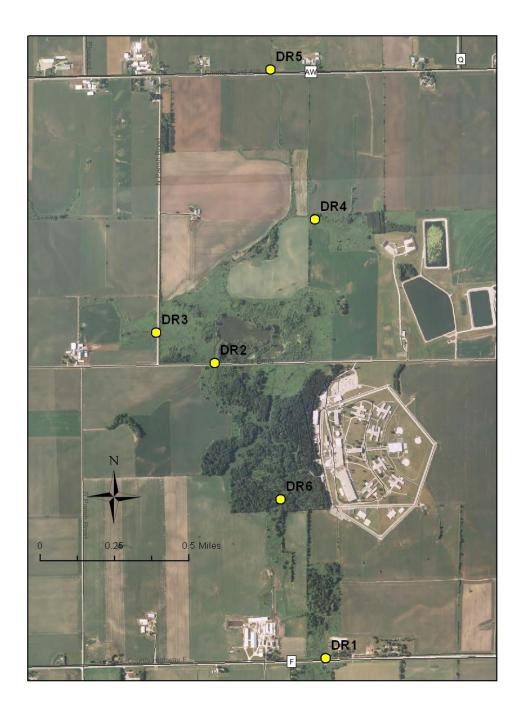


Figure 1 Location of Sampling Points

SUMMARY OF UW-MILWAUKEE 2006 SAMPLING

As discussed above, the University of Wisconsin-Milwaukee (UWM) conducted tributary monitoring of Alto, Drew and Cambra Creeks in spring of 2006. In the Drew Creek watershed samples were collected at three sites. The sites were No. 1 (CTH F), No. 2 (Lake Emily Road) and No. 3 (Parish Road) illustrated on Figure 1. The mean values of the UWM sampling are summarized in Table 1. The full results are summarized in Appendix A. Included in the table for comparisons are statewide means and ranges for nitrogen and phosphorus based on data for 240 streams as part of the study, *Nutrient Concentrations and Their Relations to the Biotic Integrity of Wadeable Streams in Wisconsin* (USGS, 2006). Values in bold text indicate mean concentrations above the state, ecoregion or environmental phosphorus zone averages as reported by USGS.

Table 1

Watershed and Station No.	Total Suspended Solids (mg/l)	Organic Nitrogen (TKN) (mg/l)	Nitrate/Nitrite (mg/l)	Total Nitrogen (mg/l)	Total Phosphorus (mg/l)	Dissolved Phosphorus (SRP) (mg/l)
Drew No. 1	8.600	0.456	18.800	19.256	0.094	0.062
Drew No. 2	10.100	0.694	21.040	21.734	0.101	0.040
Drew No. 3	71.333	1.074	20.740	21.814	0.200	0.055
		Statev	vide Means (USG	S, 2006)	-	
Mean	-	0.675	2.086	2.807	0.116	0.079
Median		0.563	1.048	1.695	0.085	0.050
Minimum	-	0.070	0.005	0.131	0.012	0.004
Maximum	-	2,350	20.550	21.260	1.641	1.495
Standard	dard -		2.865	2.860	0.144	0.122
deviation						
		Eco-re	egion Means (US	GS, 2006)		
Median	-	-	-	0.811	0.025	-
0% percentile	-	-	-	0.777	0.023	-
100 percentile	-	-	-	21.260	1.641	-
	En	vironmental Pl	hosphorus Zone	Means (USGS	S, 2006)	
Median	-	-	-	0.632	0.042	-
0% percentile	-	-	-	0.298	0.016	-
100 percentile	-	-	-	21.260	0.304	-

Results of UWM Tributary Monitoring Mean Values from Five Sampling Dates

Source: University of Wisconsin-Milwaukee and Wisconsin Laboratory of Hygiene and USGS (2006)

As can be seen Drew Creek has concentrations of organic nitrogen (TKN), nitrate/nitrite, and total phosphorus that exceed the statewide means. Concentrations of nitrate/nitrite on some dates actually exceed the USGS maximum value for their 240 watersheds making these levels some of the highest recorded in Wisconsin.

RESULTS OF HEY SAMPLING

The results of the 2008-2009 sampling by Hey and Associates are summarized in Table 2.

Site Location	Site No.	Date	Flow (cfs)	Flow (MGD)	TP (mg/l)	SRP (mg/l)	TSS (mg/l)	TKN (mg/l)	NO2 NO3 (mg/l)	E-Coli (Counts per 100 ml)
		8/5/08	10.83	7.00	0.10	0.07	7.00	0.55	15.90	866.00
		9/9/08	10.31	6.66	0.07	0.04	7.00	0.52	18.10	129.00
HWY F	DR1	10/9/08	6.89	4.45	0.09	0.05	13.00	0.67	14.50	2419.00
		7/14/09	2.79	1.80	0.07	0.04	8.00	0.27	16.70	-
		Mean			0.08	0.05	8.75	0.50	16.30	1138.00
		8/5/08	19.26	12.45	0.08	0.06	7.00	0.38	18.30	225.00
Lake		9/9/08	11.61	7.50	0.06	0.03	10.00	0.35	19.20	70.00
Emily	DR2	10/9/08	7.34	4.74	0.06	0.03	4.00	0.48	16.10	133.00
Road		7/14/09	0.78	0.50	0.07	0.03	13.00	0.15	17.80	-
		Mean			0.07	0.04	8.50	0.34	17.85	142.67
		8/5/08	3.00	1.94	0.20	0.27	3.00	0.58	13.70	196.00
	DR3	9/9/08	2.35	1.52	0.13	0.03	29.40	0.63	17.70	167.00
Parish		10/9/08	3.94	2.55	0.06	0.05	3.00	0.52	16.50	32.00
Road		7/14/09	0.62	0.40	0.08	0.04	10.00	0.30	15.30	-
		Mean			0.12	0.10	11.35	0.51	15.80	131.67
	DR4	8/5/08	3.10	2.00	0.07	0.04	22.00	0.29	19.30	121
		9/9/08	9.26	5.98	0.06	0.03	14.00	0.45	20.20	111.00
Prison		10/9/08	1.33	0.86	0.06	0.03	20.00	0.36	18.10	34.00
Creek		7/14/09	1.40	0.90	0.07	0.03	34.00	0.21	19.10	-
		Mean			0.06	0.03	22.50	0.33	19.18	88.67
		8/5/08	No Flow	No Flow	-	-	-	-	-	-
CTH AW	DR5	9/9/08	No Flow	No Flow	-	-	-	-	-	-
CILAW		10/9/08	No Flow	No Flow	-	-	-	-	-	-
		7/14/09	No Flow	No Flow	-	-	-	-	-	-
		9/9/08	9.37	6.06	0.06	0.03	8.00	0.22	18.90	88.00
Prison	000	10/9/08	2.79	1.81	0.08	0.04	16.00	0.62	15.50	517.00
Grounds	DR6	7/14/09	1.86	1.20	0.06	0.03	7.00	0.14	17.70	
		Mean			0.06	0.03	10.33	0.33	17.37	302.50

 Table 2

 Results of Drew Creek Tributary Monitoring 2008 -2009

Concentrations highlighted in Table 2 are levels above statewide means for nitrogen and phosphorus based on data for 240 streams as part of the study, *Nutrient Concentrations and Their Relations to the Biotic Integrity of Wadeable Streams in Wisconsin* (USGS, 2006), or above state water quality standards for bacteria.

Stream Flow

Stream flow was measured on each of the sampling dates. Exhibit 1 in Appendix B illustrates the variability of flow from upstream to downstream. At station DR5, at the headwater at CTH AW, no flow was observed on any of the sampling dates. At the DR3 and DR4 flows ranged from 0.4 to 9.6 cfs. At site DR2 flows increased dramatically on all of the sampling dates, indicating groundwater discharges just above this site in a large wetland complex. Much of the groundwater was likely recharged by groundwater discharges of the Fox Lake Correctional Institution Wastewater Treatment Plant which operates a series of seepage cells and a spray irrigation system. As water flowed downstream to Site DR6 and DR1 flow was lost on all of the sampling dates, likely due to water of Drew Creek seeping into the shallow bedrock that can be observed on the stream bottom.

Total Suspended Solids

Total suspended solids (TSS) include all particles suspended in water which will not pass through glass-fiber filter disk. Suspended solids are associated with nonpoint source pollution, such as soil erosion from agricultural and construction sites. As levels of TSS increase, a water body begins to lose its ability to support a diversity of aquatic life. Suspended solids absorb heat from sunlight, which increases water temperature and subsequently decreases levels of dissolved oxygen. Warmer water holds less oxygen than cooler water. Photosynthesis also decreases, since less light penetrates the water causing less oxygen to be produced by plants and algae. TSS can also destroy fish habitat because suspended solids settle to the bottom and cover coarse bottom materials. Suspended solids can smother the eggs of fish and aquatic insects, and can suffocate newly-hatched insect larvae. Suspended solids can also harm fish by clogging gills, reducing growth rates, and lowering resistance to disease. As suspended solids settle in the calm waters of the lake, they fill in bays impacting recreational use. Levels above 40 mg/l cause water to become cloudy and above 100 mg/l begin to damage aquatic life.

In 2008-2009 sampling, TSS values ranged from 3 to 34 mg/l, indicating low levels of suspended sediment. The sampling indicates that surface erosion on the sampling dates was not a major problem.

Table 3 summarizes the estimated loadings of TSS, nitrogen and phosphorus to Fox Lake at each of the Drew Creek sampling sites. Loading is the total mass in pounds that enters the lake per day or per year. Multiplying the measured stream flow times the pollutant concentration you can estimate daily pounds of a given pollutant that has entered the lake at that moment. Estimates of annual loading, as done in the 2005 -2006 UWM and Hey and Associates study would require additional flow and concentration data.

Table 3

Site	Site	Date	Flow	Flow		Lo	adings (lbs/c	lay)	
Location	No.		(cfs)	(MGD)	TP	SRP	TSS	TKN	NO2+NO3
Drew	DR1	8/5/2008	10.83	7.00	5.95	4.14	408.66	32.11	928.24
Creek at		9/9/2008	10.31	6.66	3.95	2.00	389.09	28.90	1606.61
HWY F		10/9/2008	6.89	4.45	3.34	1.74	482.56	24.87	923.17
		7/14/2009		1.80	1.01	0.54	120.10	4.05	60.85
Drew	DR2	8/5/2008	19.26	12.45	8.72	5.81	726.72	39.45	1899.85
Creek at		9/9/2008	11.61	7.50	3.63	1.69	625.79	21.90	1370.64
Lake		10/9/2008	7.34	4.74	2.45	1.35	158.25	18.99	751.28
Emily Road		7/14/2009		0.50	0.28	0.14	54.21	0.63	2.61
Drew	DR3	8/5/2008	3.00	1.94	3.18	4.40	48.50	9.38	221.46
Creek at		9/9/2008	2.35	1.52	1.61	0.43	371.84	7.97	100.78
Parish		10/9/2008	3.94	2.55	1.34	0.96	63.68	11.04	234.29
Rd.		7/14/2009		0.40	0.26	0.13	33.36	1.00	3.34
Prison	DR4	8/5/2008	3.10	2.00	1.19	0.68	367.29	4.84	322.21
Creek		9/9/2008	9.26	5.98	2.74	1.35	698.52	22.45	1120.24
		10/9/2008	1.33	0.86	0.44	0.21	142.95	2.57	18.39
		7/14/2009		0.90	0.51	0.21	255.20	1.58	11.83
Drew	DR6	9/9/2008	9.37	6.06	2.93	1.62	404.13	11.11	561.41
Creek on		10/9/2008	2.79	1.81	1.19	0.57	240.92	9.34	140.57
Prison Grounds		7/14/2009		1.20	0.55	0.33	70.06	1.40	14.02

Calculated Daily Pollutant Loadings in Pounds per day for the Drew Creek Tributary Monitoring by Sample Site 2008 -2009

Phosphorus

Aquatic plants and algae require nutrients such as phosphorus, nitrogen, carbon, calcium, chlorides, iron, magnesium, sulfur, and silica for growth. In lakes where the supply of one or more of these nutrients is limited, plant and algae growth may also be limited. The two nutrients that most often limit and control the growth of plants are nitrogen and phosphorus. In nutrient limited lakes, if you add more nitrogen or phosphorus, you will get more plant or algae growth. The Southeastern Wisconsin Regional planning Commission (SEWRPC) has recommended that in-lake total phosphorus concentration be below 0.02 mg/l to prevent nuisance algae blooms. Currently the State of Wisconsin has no surface water standards for phosphorus.

Two types of phosphorus were sampled; total phosphorus (TP) and soluble reactive phosphorus (SRP).

Phosphorus is found in natural environments in several forms including (Snoeyink and Jenkins, 1980):

- Orthophosphate (H₃PO₄, H₂PO₄, HPO₄, HPO₄², HPO₄² complexes)
- Polyphosphates $(H_4P_2O_7, H_4P_2O_7^2, H_4P_2O_7^2, H_4P_2O_7^3, H_4P_2O_7^4, H_4P_2O_7^3 \text{ complexes})$
- Metaphosphate $(HP_3O_9^2, HP_3O_9^3)$
- Organic phosphates (phosphorus tied up in organic matter)

A test for total-phosphorus (TP) will identify the combined concentration of all of the above compounds. Orthophosphate and the other complexes of phosphorus found in the natural environment are generally not very soluble and typically bind with various cations such as calcium (Ca), magnesium (Mg), aluminum (Al) or iron (Fe). The most common complexes include the following:

•	Calcium hydrogen phosphate	CaHPO4 _(s)	(pK _{so} +6.66)
•	Calcium dihydrogen phosphate	$Ca(H_2PO_4)_{2(s)}$	(pK _{so} +1.14)
•	Hydroxyapatite	$Ca_5(PO_4)_2OH_{(s)}$	(pK _{so} +55.9)
•	Ferric phosphate	FePO ₄	(pK _{so} +21.9)
•	Aluminum phosphate	AIPO ₄	(pK _{so} +21.0)

The solubility of these complexes in fresh water is defined by a solubility equilibrium constant (pK_{so}). The smaller the solubility constant the less soluble the compound is in water. Other complexes with sodium (Na), magnesium (Mg), manganese (Mn), and orthophosphate, pyrophosphate and tripophosphate also exist, making phosphorus equilibrium chemistry very complex.

The soluble reactive phosphorus (SRP) consists largely of the inorganic orthophosphate (PO_4) form of phosphorus. Orthophosphate is the phosphorus form that is directly taken up by algae, and the concentration of this fraction constitutes an index of the amount of phosphorus immediately available for algal growth. The presence of soluble phosphorus in high concentrations is unusual in areas with hard water and high calcium levels as found throughout Southern Wisconsin.

The test results show high concentration of total phosphorus on all of the sampling dates and at all of the sampling locations. The concentrations do not vary greatly by sample site, indicating that phosphorus contamination is a watershed wide problem. Exhibits 2 and 3 in Appendix B illustrate concentrations of total phosphorus and soluble reactive phosphorus by site respectively. It can be seen that approximately 60% of the total phosphorus is in the form of soluble reactive phosphorus. As outlined above this is unusual in the hard water of Southern Wisconsin. This indicates that the source of the phosphorus is not soil erosion where most phosphorus is typically bound to the soil particles, but is due a soluble source of phosphorus such as contamination of groundwater by sources such as treated wastewater, septic system waste or animal manure.

Nitrogen

Nitrogen can be found in many different organic and inorganic forms in our environment. The air we breathe is composed of 78 percent nitrogen. Nitrogen can also be found in many varied forms in the soil. Plants need nitrogen from the soil for proper growth and development but are only able to use very specific forms of nitrogen. Plants cannot use the form of nitrogen found in the atmosphere. Natural biological process carried out by microorganisms in the soil convert organic nitrogen to inorganic forms, which plants are able to use. Organic nitrogen is a common component in plant residues and organic matter. Ultimately, organic nitrogen is converted to inorganic ammonium (NH-). Nitrate is the form of nitrogen that is most used by plants for growth and development. Where crops are grown, nitrates can also emanate from nitrogen fertilizers, and manure.

Nitrogen becomes a concern to water quality when nitrogen in the soil is converted to the nitrate (NO₃) form. This is because nitrate is very mobile and easily moves with water. The concern of nitrates and water quality is generally directed at groundwater. However, nitrates can also enter surface waters such as ponds, streams and rivers. Nitrates in the soil result from natural biological processes associated with the decomposition of plant residues and organic matter. Nitrates can also come from animal manure, treated human waste effluents and nitrogen fertilizers.

Whether nitrates actually enter groundwater depends on underlying soil and/or bedrock conditions, as well as depth to groundwater. If depth to groundwater is shallow and the underlying soil is sandy, the potential for nitrates to enter groundwater is relatively high.

Two of the major problems with excess levels of nitrogen in the environment are:

- Excess nitrogen can cause overstimulation of growth of aquatic plants and algae. Excessive growth of these organisms, in turn, can clog water intakes, use up dissolved oxygen as they decompose, and block light to deeper waters. This seriously affects the respiration of fish and aquatic invertebrates, leads to a decrease in animal and plant diversity, and affects our use of the water for fishing, swimming, and boating.
- Too much nitrate in drinking water can be harmful to young infants or young livestock.

The two forms of nitrogen measured as part of this study were:

- Total Kjeldahl Nitrogen (TKN), a measurement of organic nitrogen and ammonia, and
- Nitrite/nitrate nitrogen which measures nitrogen in the forms of NO₂ and NO₃.

Total Kjeldahl Nitrogen (TKN) concentration ranged from 0.15 to 0.63 mg/l and did not show any trends from upstream to downstream. The concentrations of TKN are generally low indicating limited levels of organic nitrogen or ammonia.

Nitrite/nitrate nitrogen levels ranged from 14.5 to 20.2 mg/l. State and Federal laws set the maximum allowable level of nitrate-nitrogen in public drinking water at 10 milligrams per liter (10 parts per million). Nitrate-contaminated water should never be fed to an infant less than 6 months of age. In young infants, ingestion of nitrate can reduce the blood's ability to carry oxygen. In severe cases it can cause a condition that doctors call methemoglobinemia also called "blue baby syndrome" because the infant's skin appears blue-gray or lavender in color. This skin color change is caused by a lack of oxygen in the blood. An infant suffering from "blue baby syndrome" needs immediate medical care because the condition can lead to coma and death if it is not treated promptly. Some scientific studies have also found evidence suggesting that women who drink nitrate contaminated water during pregnancy

are more likely to have babies with birth defects. People who have heart or lung disease, certain inherited enzyme defects or cancer may be more sensitive to the toxic effects of nitrate than healthy individuals. Some researchers also suspect that consuming nitrate-contaminated water may increase the risk of certain types of cancer (WDNR Publication: PUB-DG-001 2006).

High levels of nitrate/nitrite nitrogen were found at all of the Drew Creek sampling sites. Approximately 97 % of the total nitrogen is in the form of nitrate and nitrite. Drew Creek receives most of its water from groundwater seepage, indicating that the groundwater throughout the stream's watershed is highly contaminated. Potential sources of the nitrate/nitrite nitrogen is the Fox Lake Correctional Institution wastewater treatment plant, animal feedlots and the excess spreading of manure and other organic waste. Because levels are high everywhere it is difficult to pinpoint a single specific source.

Fecal Bacteria

Fecal bacteria are bacteria that grow in the intestines of warm blooded animals and are found in fecal matter. E-coli (Escherichia coli) are a specific form of the fecal coliform group. E-coli have been associated with making humans sick through ingestion. The U. S. Environmental Protection Agency (USEPA) has recommended that E- coli be used to measure the safety of public beaches. The USEPA recommends that beaches be posted with an advisory sign informing the public of increased health risk when a water sample exceeds 235 colony-forming units of E. coli per 100 milliliters of water.

High levels of e-coli bacteria were found on one date each at stations DR2 (Lake Emily Road) and DR6 (Prison Creek), and on two dates at DR1 (HWY F). The highest values of 866 and 2,419 were found at DR1 just below the discharge of a major feedlot located to the west along HWY F.

SAMPLING BY FOX LAKE CORRECTIONAL INSTITUTION

The Fox Lake Correctional Institution (FLCI) is situated on approximately 1200 acres owned by the State of Wisconsin located in the center of the Drew Creek Watershed (Figure 2). The FLCI is operated as a medium security facility with an adjoining minimum security compound for adult male offenders. The inmate population at the prison on June 27, 2008 was 1,339 with 411 security and non-security staff. The institution operates a waste water treatment plant on the property which discharges to series of infiltration cells and into the local groundwater. Wastewater is also land applied to agricultural fields on the FLCI property through a spray irrigation system. As part of the treatment plants license it is required to monitor the local groundwater system. Four monitoring well are monitored quarterly. A map of the well locations was not provided with the data obtained from FLCI. The results of the 2007 and 2008 sampling are summarized in Table 4. Samples were collected between 10 and 39 feet below the land surface depending on the date and location.

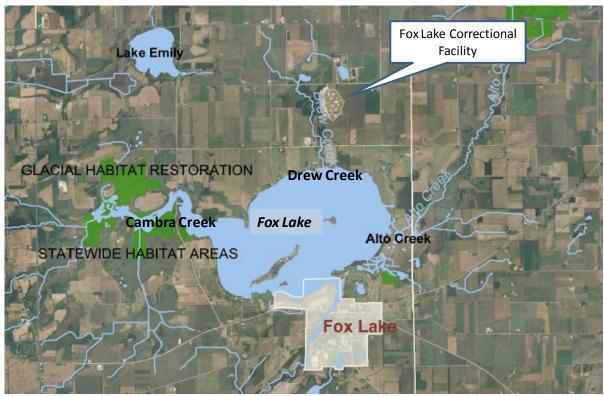


Figure 2 Location of Fox Lake Correctional Institution

As seen in Table 4, with the exception of one high value of 20.1 mg/l of nitrate/nitrite nitrogen on 9/28/2007 at one site (807), all values are below 6.10 mg/l with an average for all sites and dates of 2.5 mg/l. Levels of total Kjeldahl nitrogen, ammonia nitrogen, and dissolved organic nitrogen were low on all sampling dates. Phosphorus is not monitored by FLCI.

We know from continuous stream flow monitoring conducted by the University of Wisconsin-Milwaukee in 2004 through 2005 (Hey and Associates, Inc. and UW-Milwaukee, 2008) that the majority of flow in Drew Creek is from groundwater discharges to the creek. The stream flow should be representing the same groundwater that FLCI is monitoring. However, FLCI is measuring moderate to low levels of nitrogen and UWM and Hey and Associates are measuring very high levels of nitrate/nitrite nitrogen. This large difference in the monitored concentrations is not understood. We know from mapping of local soils (Appendix C) that bedrock is located within 5 to 20 feet below the land surface in the Drew Creek watershed. It is possible that water is percolating through the surface soil layer down to the surface of the bedrock and them moving laterally towards to stream. It is possible that FLCI is monitoring groundwater in the upper soil layer. Without more detailed mapping of the local geology it is difficult to understand the local groundwater flow patterns. A first place to start in understanding the local geology would be to examine the well logs from the FLCI monitoring wells and any other soil borings that have been conducted in the areas.

Sample Point	Date	Depth to Ground- water (feet)	рН	Chloride Dissolved (mg/l)	Total Dissolved Solids (mg/l)	Nitrite + Nitrate (mg/l)	Total Kjeldahl Nitrogen (mg/l)	Ammonia Nitrogen (mg/l)	Organic Dissolved Nitrogen (mg/l)
804	3/20/07	19.58	7.50	16	380	1.00	<0.25	<0.10	<0.25
805	3/20/07	23.79	7.49	-	430	2.50	0.34	<0.10	0.34
807	3/20/07	28.23	7.30	82	610	<0.10	<0.25	<0.10	<0.25
808	3/20/07	31.33	7.44	33	460	0.42	0.34	<0.10	0.34
804	6/12/07	19.83	7.77	7	340	0.67	0.38	<0.10	0.38
805	6/12/07	23.96	7.76	40	460	6.10	0.31	<0.10	0.31
807	6/12/07	29.57	7.38	120	590	<0.10	0.33	<0.10	0.33
808	6/12/07	34.63	7.59	23	400	0.39	0.40	<0.10	0.40
804	9/28/07	21.18	7.68	5	330	0.65	<0.25	<0.10	<0.25
805	9/28/07	23.99	7.54	44	480	4.70	0.39	<0.10	0.39
807	9/28/07	28.87	7.82	110	600	20.10	<0.25	<0.10	<0.25
808	9/28/07	32.81	7.40	67	490	0.51	<0.25	<0.10	<0.25
804	12/26/07	22.98	7.18	52	690	2.60	0.739	<0.10	0.739
805	12/26/07	25.38	7.32	48	610	3.70	0.976	<0.10	0.976
807	12/26/07	30.41	7.11	71	630	0.14	0.907	<0.10	0.907
808	12/26/07	39.92	7.05	23	500	0.20	0.441	<0.10	0.411
804	3/25/08	17.77	7.45	9	310	0.62	0.29	<0.10	0.29
805	3/25/08	23.75	8.64	63	430	2.30	0.29	<0.10	0.29
807	3/25/08	25.91	8.32	130	530	<0.10	<0.25	<0.10	<0.25
808	3/25/08	24.95	9.13	45	390	0.66	0.27	<0.10	0.27
804	6/24/08	10.07	7.52	20	340	1.00	<0.25	<0.10	<0.25
805	6/24/08	17.06	7.30	110	580	2.90	2.90	<0.10	<0.25
807	6/24/08	18.37	7.45	98	600	<0.10	<0.25	<0.10	<0.25
808	6/24/08	12.82	8.02	120	580	2.60	<0.25	<0.10	<0.25
804	9/16/08	19.39	7.50	65	570	2.10	<0.25	<0.10	<0.25
805	9/16/08	24.08	7.58	79	560	4.40	<0.25	<0.10	<0.25
807	9/16/08	29.44	7.34	130	630	<0.10	<0.25	<0.10	<0.25
808	9/16/08	35.12	7.46	51	460	0.59	<0.25	<0.10	<0.25
804	12/30/08	22.77	7.16	86	560	1.90	<0.25	<0.10	<0.25
805	12/30/08	24.50	7.30	54	460	2.20	<0.25	<0.10	<0.25
807	12/30/08	30.13	7.27	110	540	<0.10	<0.25	<0.10	<0.25
808	12/30/08	38.41	7.32	35	390	0.23	<0.25	<0.10	<0.25

 Table 4

 Groundwater Sampling by Fox Lake Correctional Institution

CONCLUSIONS

It's been stated that "A lake is a Reflection of its Watershed". In the case of Fox Lake that reflection is one of high levels of pollution. Based on monitoring by the University of Wisconsin-Milwaukee in 2004 and 2005 it was estimated that annually approximately 499,000 pounds of sediment, 430,690 pounds of nitrogen and 44,330 pounds of phosphorus entered Fox Lake from its watershed. To identify the sources of the sediment and nutrients entering Fox Lake a sampling program of the Drew Creek watershed was conducted in 2008 and 2009.

The most shocking result of the sampling was the high levels of nitrate/nitrite nitrogen (NO_2+NO_3) found at all of the sampling sites. Nitrite/nitrate nitrogen levels ranged from 14.5 to 20.2 mg/l well above the state's drinking water standard of 10 mg/l.

Sediment levels in the water were generally low ranging from 3.0 to 34.0 mg/l. The largest source of sediment is runoff from agricultural fields. The Drew Creek watershed contains soils with moderate levels of sand and allows rainwater to seep into the ground during moderate storms. To generate large enough runoff events to move sediment we need rain storms with high intensity or on frozen ground. None of the sampling took place during one of these large storms. This does not mean that Drew Creek is not a source of sediment to the lake.

While the State of Wisconsin does not have water quality standards for phosphorus, scientists believe levels above 0.02 mg/l can cause nuisance algae blooms. All of the monitored values were above 0.06 mg/l, three times above what is considered safe for the lake.

E-coli bacteria (Escherichia coli) are bacteria that live in the digestive track of warm blooded animals including man and livestock. The presence of e-coli bacteria in the water is an indication of animal waste. To protect public health the U. S. Environmental Protection Agency has recommended that beaches be closed when e-coli levels exceed 235 CFU/100 ml. In Drew Creek levels at CTH F exceeded this value on more than half of the sampling dates, indicating an upstream source of animal waste, likely a local dairy farm.

RECOMMENDATIONS FOR FUTURE ACTION

- E-coli bacteria sampling indicates the Shultz farm on CTH F and Drew farm on Lake Emily Road may be sources of animal waste entering Drew Creek and Fox Lake. The Wisconsin Department of Natural Resources should conduct an investigation under Wisconsin Administrative Code NR243 as to whether or not these farms should be ordered to take corrective actions.
- The Drew Creek watershed has been identified as an area with high potential for groundwater contamination by the U. S. Geological Survey (Appendix C). Nitrate/nitrate and phosphorus sampling of Drew Creek has demonstrated that this contamination has already taken place. Likely sources of the contamination include barnyard runoff, spreading of manure and other industrial waste, and land

application of treated wastewater from the Fox Lake Correctional Institution. Based on the existing problems three things should take place:

- a. No new sources of animal or human waste should be imported into the watershed.
- b. Existing land applications of animal or human waste should be minimized. The Wisconsin Legislature should explore new regulations for the land application of waste in areas with high potential for groundwater contamination.
- c. The Fox Lake Correctional Institution (FLCI), Wisconsin Department Natural Resources (WDNR) and Fox Lake Inland Lake Protection and Rehabilitation District should explore the development of treatment system to remove nitrogen and phosphorus from the Drew Creek flow before it enters the lake. The system would need to be designed to remove soluble nitrogen and soluble phosphorus. Practices such as wet detention designed to remove particulate pollutants will not be effective on these soluble pollutants. Potential treatment systems could include artificial wetlands treatment and or alum treatment systems.

REFERENCES

- Edwards, T.K., and G.D. Glysson. (1999). Field Methods for Measurement of Fluvial Sediment, Book 3, Chapter C2. Techniques of Water-Resources Investigations of the United States Geological Survey, U.S. Government Printing Office, Washington, DC.
- Hey and Associates, Inc. and UW-Milwaukee (2008). Fox Lake Management Strategy Evaluation and Recommendations for Future Action – 2008, Brookfield, WI.
- R. A. Smith and Associates, Inc. (1998). Long Range Planning Strategy for the Rehabilitation of Fox Lake, Dodge County, Brookfield, WI
- Robertson, D.M., Graczyk, D.J., Garrison, P.J., Wang, L., LaLiberte, G., and Bannerman, R., (2006). Nutrient concentrations and their relations to the biotic integrity of wadeable streams in Wisconsin: U.S. Geological Survey Professional Paper 1722, 139 p. Madison, WI.
- Shelton, L. R., (1994). Field Guide for Collecting and Processing Stream Water Samples for the National Water-Quality Assessment Program, Open-File Report 94-455, United States Geological Survey, Sacramento, California.
- United States Geological Survey (USGS). (2005). Techniques of Water Resources Investigations Reports. Book 3: Applications of hydraulics, Section A: Surface-water techniques. (21 chapters). United States Department of Interior, U.S. Geological Survey. Washington D.C. http://water.usgs.gov/pubs/twri/.